

4 (a) Show for helium that $\gamma = 5/3$ where γ is the adiabatic constant.

A closed-cycle single-shaft gas turbine plant using helium as the working fluid incorporates the following components in the given order: (a) a compressor, (b) a heater, (c) a two-stage turbine with reheater and (d) a cooler.

The maximum and minimum pressures and temperatures in the cycle are 40 bar and 700 °C, and 10 bar and 25 °C respectively, with reheat to 700 °C. The pressure in the reheater is optimum for maximum specific power (power per kg/s of gas flow).

The molar mass of helium is 4 kg/kmol and the molar heat capacity at constant volume for helium is $3/2 \tilde{R}$ where $\tilde{R} = 8.3145$ kJ/kmol K is the universal molar gas constant.

- (b) Sketch the T-s diagram for the plant and indicate pressures and temperatures between the components if
- (i) the reheater is used,
 - (ii) the reheater is by-passed.

(c) Calculate the ideal cycle efficiency and specific power for each case. Assume that there are no losses in the cycle.

(a) For Helium $\tilde{m} = 4$ (mol mass) $R = \tilde{R} / \tilde{m} = 8.3145/4 = 2.0786$ kJ/kg K

$$\tilde{c}_v = \frac{3\tilde{R}}{2}$$

$$\tilde{c}_p = \tilde{R} + \tilde{c}_v = \tilde{R} + \frac{3\tilde{R}}{2} = \frac{5\tilde{R}}{2}$$

$$c_v = \frac{3R}{2} = 3.1179 \text{ kJ/kg K}$$

$$c_p = R + c_v = 5.1966 \text{ kJ/kg K}$$

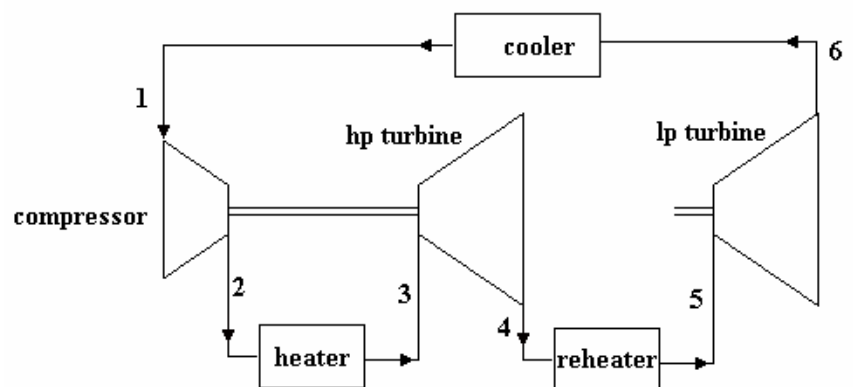
$$\gamma = c_p / c_v = 1.667$$

$$p_1 = 10 \text{ bar} \quad \theta_1 = 25^\circ\text{C} \quad T_1 = 298 \text{ K}$$

$$p_2 = 40 \text{ bar}$$

$$p_3 = 40 \text{ bar} \quad \theta_3 = 700^\circ\text{C} \quad T_3 = 973 \text{ K}$$

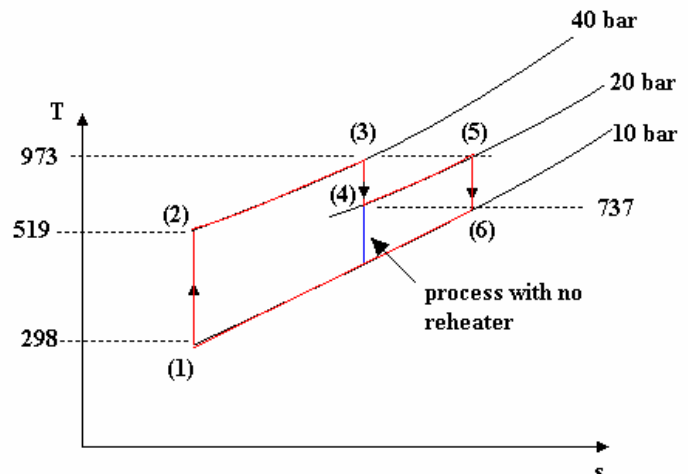
$$\text{For optimal turbine work } p_{4/5} = \sqrt{(40)(10)} = \sqrt{400} = 20 \text{ bar} \quad \theta_5 = 700^\circ\text{C} \quad T_5 = 973 \text{ K}$$



$$T_2 = T_1 \left(\frac{p_2}{p_1} \right)^{\frac{1-\gamma}{\gamma}} = 298 \left(\frac{40}{10} \right)^{\frac{1-1.667}{1.667}} = 518.9 \text{ K}$$

$$T_4 = T_3 \left(\frac{p_4}{p_3} \right)^{\frac{1-\gamma}{\gamma}} = 973 \left(\frac{20}{40} \right)^{\frac{1-1.667}{1.667}} = 737.4 \text{ K}$$

$$T_6 = T_5 \left(\frac{p_6}{p_5} \right)^{\frac{1-\gamma}{\gamma}} = 973 \left(\frac{10}{20} \right)^{\frac{1-1.667}{1.667}} = 737.4 \text{ K}$$



HEAT INPUT

$$\Phi(\text{in}) = c_p(T_3 - T_2) + c_p(T_5 - T_4) = 5.1966(973 - 518.9) + 5.1966(973 - 734.7) = 3598.1 \text{ kW}$$

HEAT OUTPUT

$$\Phi(\text{out}) = c_p(T_6 - T_1) = 5.1966(734.7 - 298) = 2269.4 \text{ kW}$$

$$\text{Nett Power Out} = 3598.1 - 2269.4 = 1328.7 \text{ kW per kg/s of gas flow}$$

$$\text{Cycle efficiency } \eta = P/\Phi(\text{in}) = 1328.7/3598.1 = 0.369 \text{ or } 36.9 \% \text{ with reheater}$$

With the reheater bypassed we have a standard Joule cycle.

$$\eta = 1 - r_p^{\frac{1}{\gamma} - 1} = 1 - \left(\frac{40}{10}\right)^{\frac{1}{1.667} - 1} = 0.426$$

HEAT INPUT

$$\Phi(\text{in}) = c_p(T_3 - T_2) = 5.1966(973 - 518.9) = 2360 \text{ kW}$$

$$\text{Nett Power Out} = \eta \times 2360 = 1005 \text{ kW per kg/s of gas flow}$$