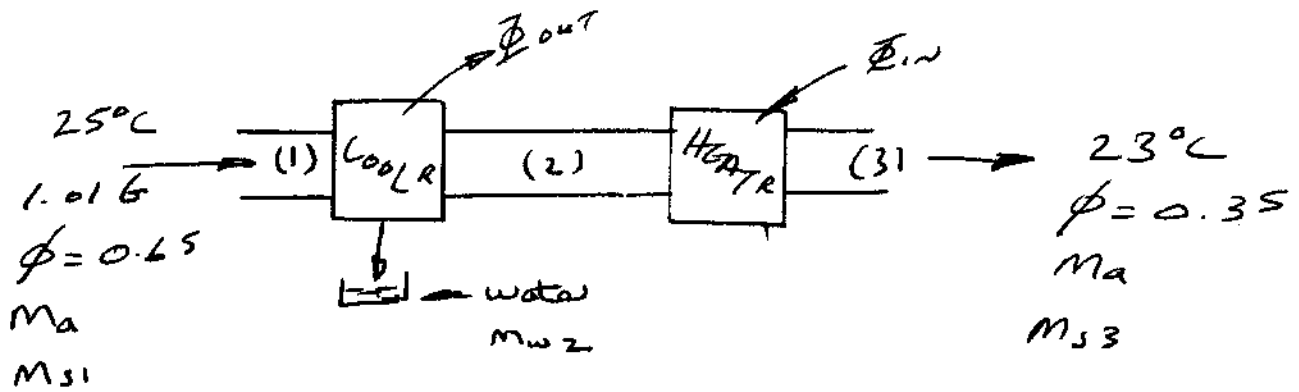


Q8 2002



M_a = MASS OF ^{dry} AIR - CONSTANT THROUGHOUT
 M_s = MASS OF VAPOUR M_w = MASS OF WATER

$$p_g = 0.03166 \text{ bar @ } 25^\circ\text{C} \quad p_s = \phi p_g$$

$$p_{s1} = 0.65 \times 0.03166 = 0.020579 \text{ bar}$$

$$p_a = 1.01 - p_{s1} = 0.989421 \text{ bar}$$

$$\omega_1 = 0.622 \frac{p_{s1}}{p_a} = 0.012937$$

$$M_a = \frac{p V}{RT} = \frac{0.989421 \times 10^5 \times V_A}{287 \times 298}$$

For 1 kg of DRY AIR $V_A = 0.864 \text{ m}^3$

$$M_{s1} = \frac{p_{s1} V}{RT} \quad \text{FOR VAPOUR VOLUME OF VAPOUR IS SAME AS VOL OF AIR}$$

$R = 462$ FOR WATER VAPOUR

$$M_{s1} = \frac{0.020579 \times 10^5 \times 0.864}{462 \times 298} = 0.0129266 \text{ kg}$$

$$\omega = M_s / M_a = 0.622 p_s / p_a$$

$$\phi = p_s / p_g = 1.608 p_a / p_g$$

$$p = 1.016 \quad \theta = 25^\circ\text{C} \quad \phi = 0.65$$

INLET

$$@ 25^{\circ}\text{C} \quad p_g = 0.03166 \text{ bar}$$

$$\phi = p_s/p_g = 0.65 \quad p_s = 0.65 \times 0.03166 \\ = 0.020579 \text{ bar}$$

$$p_a = 1.01 - 0.020579 = 0.989421 \text{ bar}$$

$$\omega = 0.622 \times \frac{0.020579}{0.989421} = 0.012937$$

$$\text{DEW POINT} = 17.8^{\circ}\text{C}$$

EXIT

$$\phi_3 = 0.35 = p_{s3}/p_{g3}$$

$$p_{g3} = p_s @ 23^{\circ}\text{C} = 0.02808 \text{ bar}$$

$$p_{s3} = 0.02808 \times 0.35 = 0.009828 \text{ bar}$$

$$p_{a3} = 1.01 - 0.009828 = 1.000172 \text{ bar}$$

$$\omega_3 = 0.622 p_s/p_a = 0.622 \times \frac{0.009828}{1.000172}$$

$$\omega_3 = 0.00611964$$

$$M_{s3} = 0.006112 \text{ Ma}$$

$$M_{s1} = 0.0129206 \text{ Ma}$$

$$\text{CONDENSATE FORMED} = M_{s1} - M_{s3} = 0.00681 \text{ Ma}$$

ENERGY BALANCE ON COOLER

$$p_{s2} = p_{s3} = 0.009828 \text{ bar}$$

$$M_a C_a (T_{a1} - T_{a2}) - M_w C_w T_w + M_{s1} h_{s1}$$

$$- M_{s2} h_{s2} = \dot{Q}_{\text{out}}$$

$$h_{s1} @ 25^{\circ}\text{C} \quad 0.0206 \text{ bar} = 2550 \text{ kJ/kg (Chart)}$$

$$h_{s2} @ 17.8^{\circ}\text{C} \quad 0.00983 \text{ bar} = h_g = 2533 \text{ kJ/kg}$$

$$M_{a1} = 1 \text{ kg} \quad C_a = 1.004 \text{ kJ/kgK}$$

$$M_{s2} = M_{s3} \quad C_w = 4.186 \text{ kJ/kgK}$$

$$1 \times 1.004 (25 - 17.8) - 0.00681 \times 4.186 \times 17.8$$

$$+ 0.0129206 \times 2550 - 0.006112 \times 2533 = \dot{Q}_{\text{out}}$$

$$\dot{Q}_{\text{out}} = 24.187 \text{ kJ for 1 kg of DRY AIR}$$

ENERGY BALANCE ON HEATER

$$h_{s3} = 2545 \text{ kJ/kg} \quad (23^\circ\text{C} \quad 0.0098286)$$

$$M_a C_a \theta_3 + M_{s3} h_{s3} = M_a C_a \theta_2 + M_{s2} h_{s2} + \dot{Q} \text{ (in)}$$

$$1 \times 1.004 \times 23 + 0.006112 \times 2545$$

$$= 1 \times 1.004 \times 17.8 + 0.006112 \times 2533 + \dot{Q} \text{ (in)}$$

$$\dot{Q} \text{ (in)} = 5.3 \text{ kJ per kg of DRY AIR}$$

$$M_a = 1 \text{ kg} \quad \text{AT EXIT} \quad M = M_a + M_{s3}$$

$$M = 1.006112 \text{ kg}$$

$$\dot{Q} \text{ (out)} = 24.04 \text{ kJ/kg}$$

$$\dot{Q} \text{ (in)} = 5.27 \text{ kJ/kg}$$

} PER kg OF
CONDITIONED
AIR