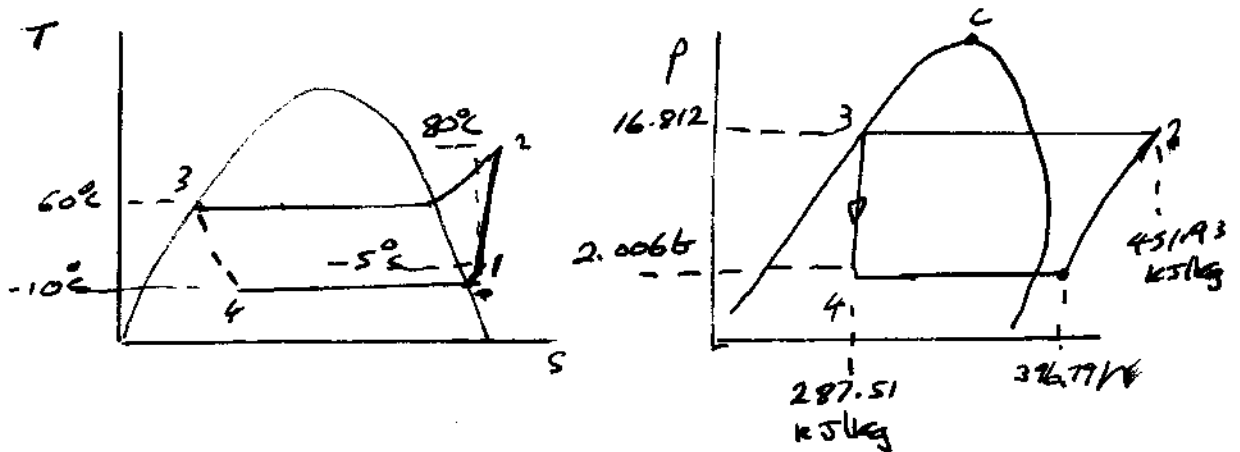
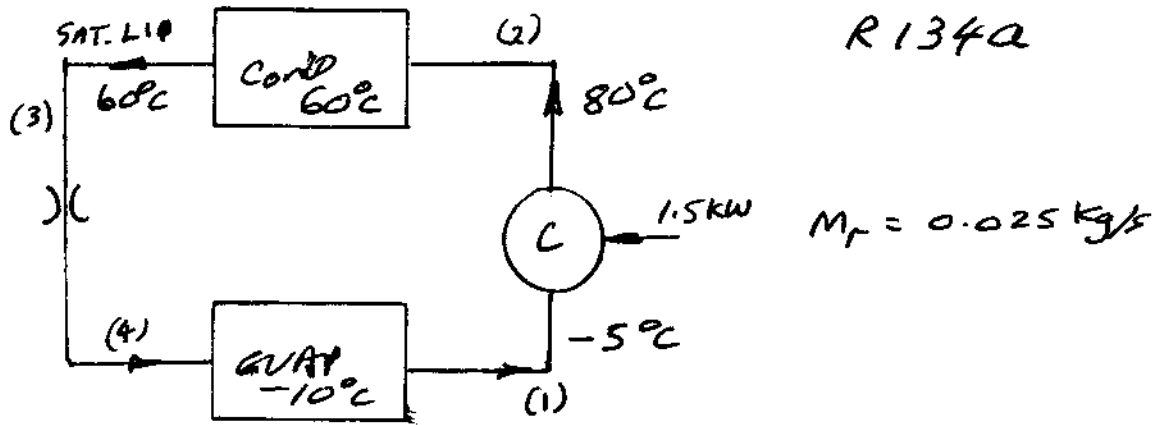


Q1 2001



To find h_1

AT 2.006 bar	-10°C	$h = 392.51$
	-5°C	Mid Point $h = 396.79$
10k Superheat	0°C	$h = 406.07$

h_2 @ 16.812 bar $t_s = 60^\circ\text{C}$
 $\therefore 80^\circ\text{C}$ is 20k Superheat

h_2 @ 16.812 bar 20k s.h. is 451.93 kJ/kg

$h_3 = h_f$ @ 16.812 bar = 287.51 kJ/kg

$$\text{HEAT PUMP} \quad \text{COP} = \frac{\dot{Q}(\text{out})}{P(\text{in})}$$

$$\begin{aligned}\dot{Q}(\text{out}) &= m_r (h_2 - h_3) \\ &= 0.025 (451.93 - 287.51) \\ &= 4.11 \text{ kW}\end{aligned}$$

$$\text{COP} = 4.11 / 1.5 = \underline{\underline{2.74}}$$

POWER PASSED INTO THE REFRIGERANT
IS $m_r (h_2 - h_1)$

$$\begin{aligned}&= 0.025 (451.93 - 396.79) \\ &= 1.3785 \text{ kW}\end{aligned}$$

$$\begin{aligned}\text{POWER LOSS FROM CASING} &= 1.5 - 1.3785 \\ &= \underline{\underline{0.1215 \text{ kW}}}\end{aligned}$$

IF COOLED TO 55°C AT POINT (3)

$h_3 \approx h_f @ 55^\circ\text{C}$ (NEAREST WE CAN GET)

$$h_3 \approx 279.46 \text{ kJ/kg}$$

$$\dot{Q}(\text{out}) = 0.025 (451.93 - 279.46) = 4.311 \text{ kW}$$

$$\text{COP} = 4.311 / 1.5 = 2.874$$

AN IMPROVEMENT AS EXPECTED BUT
WE WOULD NEED MORE EVAPORATION
TO MAINTAIN SATURATED CONDITIONS.